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B.Sc. (Bio-Technology) 6th Semester (New Scheme)

Examination, July-2022

BIO PROCESS TECHNOLOGY

Paper-BT-603

Time allowed : 3 hours] [Maximum marks : 40

Instructions :

Each question carries equal marks.

Question No. 1 is compulsory.

Attempt total five questions choosing one from each unit.

1. Define following :

(i) Fed batch culture

(ii) Baffles

(iii) Mass transfer coefficient

(iv) Single cell protein

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Unit-I

2. Describe Monod's model of substrate limited growth. 8
3. Write notes on Continuous Cultivation. 8

Unit-II

4. Write notes on Air lift reactor. 8
5. Give an account of different components of microbial medium discussing their role and common sources. 8

Unit-III

6. What do you understand by mass transfer in bioreactor ? Discuss the oxygen transfer in bioreactor. 8

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7. Give an account of different types of biosensors used in bioreactor. 8

Unit-IV

8. Write notes on microbial production of Lactic acid. 8
9. Describe the process of effluent treatment using microbes. 8

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B.Sc. 6th Semester (New Scheme) Examination,

July-2022

BIO-TECHNOLOGY

Paper-BT-602

Animal Biotechnology

Time allowed : 3 hours] [Maximum marks : 40

Instruction : Question No. 1 is compulsory. Students should attempt four other questions, i.e., one from each unit.

1. Write a short note on the following : 5×2

(a) Retroviral gene transfer

(b) Trypanosomiasis

(c) Adult stem cells

(d) Germline gene therapy

(e) Transgenic sheep

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Unit-I

2. Write a note on microinjection mediated gene transfer in animals. 7.5
3. Write a note on embryonic stem cells-mediated gene transfer in animals. 7.5

Unit-II

4. Write a note on Foot and Mouth Disease in relation to Biotechnology. 7.5
5. What do you mean by transgenesis ? Discuss with the examples. 7.5

Unit-III

6. Write a note on artificial insemination techniques for animals. 7.5

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7. Write a note on various techniques of embryo transfer. 7.5

Unit-IV

8. What do you mean by gene therapy and discuss its types and applications. 7.5
9. Write a note on molecular engineering and discuss problems associated with it. 7.5

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- (b) Write a note on electrophoresis. 2
8. (a) Describe the primary and secondary structure of proteins. 4
- (b) Explain N-terminal residue analysis of polypeptides. 4

B.Sc. 6th Semester New Scheme Examination,**July-2022****BIO-TECHNOLOGY****Paper-BT-606/BIN-606****Organic Chemistry***Time allowed : 3 hours]**[Maximum marks : 40*

Note : Attempt five questions in all, selecting at least two questions from each section.

Section-A

1. (a) Describe the Paal-Knorr synthesis for pyrrole and furan derivatives. 4
- (b) Compare the reactivity of pyrrole and pyridine towards electrophilic substitution reactions. 4
2. (a) Out of pyrrole and thiophene, which one is more aromatic and why? 3

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- (b) Explain the mechanism of nucleophilic substitution reactions in pyridine derivatives. 3
- (c) Out of pyridine and pyrrole, which one is more basic and why? 2
3. Describe with mechanism : 4,4
- (i) Bischler-Napieralski synthesis
- (ii) Fischer Indole synthesis 4
4. (a) Describe : 4
- (i) Methods of preparation of thiols
- (ii) Cleansing action of detergents
- (b) Write the preparation and use of : 4
- (i) Sulphaguanidine
- (ii) Sulphonamides

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Section-B

5. (a) What are enolates ? Describe the mechanism of Claisen Condensation. 4
- (b) Explain the keto enol tautomerism in ethyl acetoacetate. 4
6. (a) Write a note on natural and synthetic rubber. 2
- (b) Describe : 6
- (i) Phenol formaldehyde resins
- (ii) Urea formaldehyde resins
- (iii) Dacron
7. (a) Describe with examples : 6
- (i) Acidic and basic amino acid
- (ii) Fibrous and globular proteins

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9. Write the notes on any two :

2×4=8

- (a) Different levels of bio-safety
- (b) Elements of GMP
- (c) Role of GLP in Bio-safety practice

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B.Sc. 6th Semester (New Scheme) Examination,

July-2022

BIO-TECHNOLOGY

Paper-BT-601

I.P.R., Entrepreneurship, Bio-Ethics and Bio-Safety

Time allowed : 3 hours] [Maximum marks : 40

Note : Question No. 1 is compulsory and attempt four more questions by selecting one question from each unit. All questions carry equal marks.

1. Write the short notes on following : 8×1=8

- (a) Exclusive Marketing Rights
- (b) WIPO
- (c) Buffer stock
- (d) Raw material
- (e) Euthanasia

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(f) Ethical Committee

(g) Heath hazardous

(h) Bioterrorism agent

Unit-1

2. What is the Patent ? Describe the role of patenting in the field of biotechnology ? 8

3. Write the notes on any two : 2×4=8

(a) IPR & WTO

(b) Various Rights of Patentee

(c) Procedure of Patenting

Unit-2

4. Describe the role of selection, design and development of a product in the Entrepreneurship ? 8

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5. Write the notes on any two : 2×4=8

(a) Basic regulation of excise duty

(b) Elements of good entrepreneurship

(c) Effect of financial capability of entrepreneurship

Unit-3

6. Describe the roles of bio-ethics in the medical profession in detail. 8

7. Requirement of bio-ethics at national and international level. 8

Unit-4

8. What is Bio-safety ? Why are we required to follow the bio-safety rules ? 8

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- (b) Silicones and phosphazenes are isoelectronic. Discuss consequences. 4
9. (a) Draw polymeric backbone of Silicones. 4
- (b) What are Silicones ? How are cross linked Silicones prepared ? 4

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BIO-TECHNOLOGY

Paper-BT-607/BIN-605

Inorganic Chemistry

Time allowed : 3 hours] [Maximum marks : 40

Note : Attempt five questions in all, selecting one question from each section. Q. No. 1 is compulsory.

1. (a) Write IUPAC name of
 $[(CO)_3CO(CO)_2CO(CO)_3]$ 1
- (b) Give two examples of π -acid ligands similar to CO. 1
- (c) Give Lux-Flood concept of acids and bases. 1
- (d) What is HSAB principle ? 1
- (e) Give names of three essential elements in biological process. 1

94177-P-4-Q-9 (22)

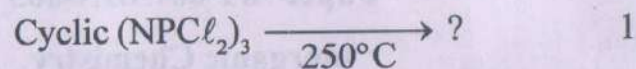
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(f) Define Nitrogen fixation. 1

(g) Complete the reaction



(h) Write common structural unit of all phosphazenes. 1

Section-A

2. Write preparation, properties and bonding of organo aluminium compounds. 8

3. Describe the nature of bonding in linear metal carbonyls and also give evidence in support of bonding. 8

Section-B

4. (a) What is symbiosis? Discuss theoretical basis of hardness and softness. 4

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(b) Explain how electronegativity can be used to explain hardness and softness of acids and bases. 4

5. What is solvent system concept of acids and bases? Write application of this concept. Also give limitation of this concept. 8

Section-C

6. What are metalloporphyrins? Discuss the structure and role played by myoglobin as O₂ carriers. 8

7. (i) Discuss biological role of Calcium ions. 4

(ii) Explain the mechanism of Na⁺ - K⁺ pump. 4

Section-D

8. (a) Write a brief account of

(i) Silicone rubbers

(ii) Silicone resins 4

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7. (a) Explain the following terms :
- (i) Phase
 - (ii) Component
 - (iii) Degree of freedom 3
- (b) Discuss Phase diagram of one component sulphur system. 5
8. (a) What is condensed system ? Explain the phase rule for condensed systems. 3
- (b) Explain the following terms
- (i) Stable equilibrium
 - (ii) Metastable state
 - (iii) Eutectic Point 3
- (c) Discuss phase equilibria involving desilverisation of Lead. 2

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B.Sc. (Bio-Technology) 6th Semester (New Scheme)

Examination, July-2022

PHYSICAL CHEMISTRY

Paper-BT-605/BIN-604

Time allowed : 3 hours]

[Maximum marks : 40

Note : Attempt five questions in all, selecting at least two questions from each section.

Section-A

1. (a) State and explain Frank-Condon principle with diagram. 4
- (b) Discuss molecular orbital diagram of carbonyl group and explain the electronic transitions taking place between them. 4
2. (a) What do you mean by multiplicity of state ? Explain with a suitable example. 3

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- (b) State and explain Stark-Einstein law of photochemical equivalence. 3
- (c) Define the term one Einstein ? How it is calculated in KJ/mol ? 2
3. (a) Write a short note on photosensitization and quenching. 3
- (b) What is Born-Oppenheimer approximation ? Explain ? 3
- (c) What do you mean by photochemical reactions ? Give at least two examples. 2
4. (a) Explain the following terms
- (i) Quantum yield
 - (ii) Non-radiative processes
 - (iii) Fluorescence
 - (iv) Phosphorescence 4

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- (b) Discuss energy transfer processes in photosensitized reactions. 4

Section-B

5. (a) Explain the terms molarity, molality, normality and mole fraction. 2
- (b) Explain Ideal and Non-ideal solutions with two example each. 3
- (c) 0.1 M solution of KNO_3 has an osmotic pressure of 4.5 atmosphere at 300 K. Calculate the apparent degree of dissociation of the salt. 3
6. (a) What are term Azeotropes ? Explain the types of azeotropes. 4
- (b) Explain osmosis law of osmotic pressure. Explain determination of molecular weight from osmotic pressure. 4

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9. Describe in detail

- (a) Ores enrichment by microorganisms 4
- (b) Control of other insects swarming the agricultural fields using biological means. 4

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PLANT BIOTECHNOLOGY AND
ENVIRONMENTAL BIOTECHNOLOGY

Paper-BT-604

Time allowed : 3 hours] [Maximum marks : 40

Note : Attempt five questions in total one from each unit.

Question No. 1 is compulsory.

1. Write short note on the following :

- (a) Micropropagation
- (b) Axillary bud
- (c) Somatic hybrids
- (d) Protoplast fusion
- (e) Bioremediation
- (f) Xenobiotic compounds

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(g) Biofertilizers

(h) *Bacillus thuringiensis*

8×1

Unit-I

2. Write a detailed description on

(a) Meristem culture

3

(b) Somaclonal variations and its applications

5

3. What is embryogenesis in tissue culture? Give a detailed description of its types and its applications? 8

Unit-II

4. What is the principle of protoplast isolation? Explain in detail its steps and application in plant biotechnology. 8

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5. Give a detailed account on

(a) Hybrids and cybrids

4

(b) Endosperm culture

4

Unit-III

6. Write a detailed account on microbiological quality control of food and water. 8

7. Discuss in detail the treatment of municipal waste and industrial effluents using microbiology approach. 8

Unit-IV

8. Write a note on

(a) Nitrogen fixation by microorganism and its importance 4

(b) Natural pesticides 4

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